

1

2

3

4

5

6

7

0

## Key

relative atomic mass
<b>atomic symbol</b>
name
atomic (proton) number

1
<b>H</b>
hydrogen
1

4
<b>He</b>
helium
2

7 <b>Li</b> lithium 3	9 <b>Be</b> beryllium 4
23 <b>Na</b> sodium 11	24 <b>Mg</b> magnesium 12

11 <b>B</b> boron 5	12 <b>C</b> carbon 6	14 <b>N</b> nitrogen 7	16 <b>O</b> oxygen 8	19 <b>F</b> fluorine 9	20 <b>Ne</b> neon 10
27 <b>Al</b> aluminium 13	28 <b>Si</b> silicon 14	31 <b>P</b> phosphorus 15	32 <b>S</b> sulfur 16	35.5 <b>Cl</b> chlorine 17	40 <b>Ar</b> argon 18

39 <b>K</b> potassium 19	40 <b>Ca</b> calcium 20	45 <b>Sc</b> scandium 21	48 <b>Ti</b> titanium 22	51 <b>V</b> vanadium 23	52 <b>Cr</b> chromium 24	55 <b>Mn</b> manganese 25	56 <b>Fe</b> iron 26	59 <b>Co</b> cobalt 27	59 <b>Ni</b> nickel 28	63.5 <b>Cu</b> copper 29	65 <b>Zn</b> zinc 30	70 <b>Ga</b> gallium 31	73 <b>Ge</b> germanium 32	75 <b>As</b> arsenic 33	79 <b>Se</b> selenium 34	80 <b>Br</b> bromine 35	84 <b>Kr</b> krypton 36
85 <b>Rb</b> rubidium 37	88 <b>Sr</b> strontium 38	89 <b>Y</b> yttrium 39	91 <b>Zr</b> zirconium 40	93 <b>Nb</b> niobium 41	96 <b>Mo</b> molybdenum 42	[98] <b>Tc</b> technetium 43	101 <b>Ru</b> ruthenium 44	103 <b>Rh</b> rhodium 45	106 <b>Pd</b> palladium 46	108 <b>Ag</b> silver 47	112 <b>Cd</b> cadmium 48	115 <b>In</b> indium 49	119 <b>Sn</b> tin 50	122 <b>Sb</b> antimony 51	128 <b>Te</b> tellurium 52	127 <b>I</b> iodine 53	131 <b>Xe</b> xenon 54
133 <b>Cs</b> caesium 55	137 <b>Ba</b> barium 56	139 <b>La*</b> lanthanum 57	178 <b>Hf</b> hafnium 72	181 <b>Ta</b> tantalum 73	184 <b>W</b> tungsten 74	186 <b>Re</b> rhenium 75	190 <b>Os</b> osmium 76	192 <b>Ir</b> iridium 77	195 <b>Pt</b> platinum 78	197 <b>Au</b> gold 79	201 <b>Hg</b> mercury 80	204 <b>Tl</b> thallium 81	207 <b>Pb</b> lead 82	209 <b>Bi</b> bismuth 83	[209] <b>Po</b> polonium 84	[210] <b>At</b> astatine 85	[222] <b>Rn</b> radon 86
[223] <b>Fr</b> francium 87	[226] <b>Ra</b> radium 88	[227] <b>Ac*</b> actinium 89	[261] <b>Rf</b> rutherfordium 104	[262] <b>Db</b> dubnium 105	[266] <b>Sg</b> seaborgium 106	[264] <b>Bh</b> bohrium 107	[277] <b>Hs</b> hassium 108	[268] <b>Mt</b> meitnerium 109	[271] <b>Ds</b> darmstadtium 110	[272] <b>Rg</b> roentgenium 111	Elements with atomic numbers 112 – 116 have been reported but not fully authenticated						

\* The Lanthanides (atomic numbers 58 – 71) and the Actinides (atomic numbers 90 – 103) have been omitted.

Relative atomic masses for **Cu** and **Cl** have not been rounded to the nearest whole number.

# Forces

## Key Vocabulary:

**Acceleration:** The rate at which an object's velocity changes

**Air resistance:** The force of air acting on a moving object

**Balanced forces:** Two forces of equal size acting in opposite directions

**Contact force:** A force that must touch an object to affect it

**Friction:** The force caused by one surface touching another surface

**Gravity:** A force that attracts an object towards the centre of another object

**Magnetism:** The force between two magnets or between a magnet and a magnetic material

**Motion:** Movement

**Newton:** The unit for force

**Non-contact force:** A force that can affect an object without touching it

**Tension:** The force acting on an object that has been stretched

**Thrust:** A 'pushing' force

**Up-thrust:** The force that acts upwards on an object, often from air-resistance or water

**Velocity:** The scientific word for 'speed'

**Weight:** The force that results from an object's mass and the effect of gravity

## Life.

**7 life processes: (MRS GREN).** Movement, Respiration, Sensitivity, Growth, Reproduction, Excretion, Nutrition.

**Habitat:** Is where an organism lives, it contains everything the organism needs to survive.

**Ecology:** the relations of organisms to one another and to their physical surroundings.

**Environment:** everything within the surroundings of a specific area.

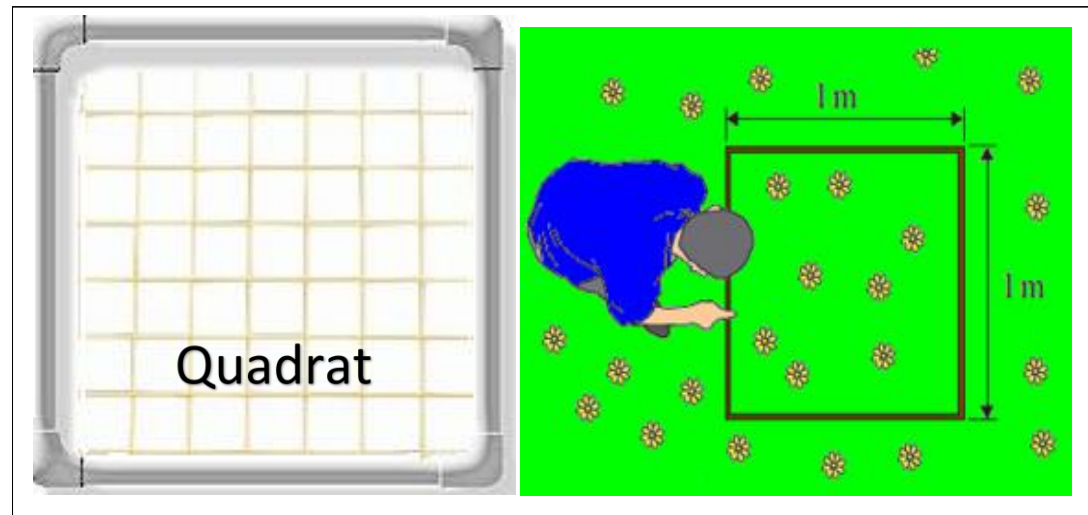
**Sample:** A small part or quantity intended to show what the whole is like.

**Population:** the number of organisms of the same species within a certain area

**Abundance:** A measure of how common or rare something is.

**Distribution:** Where particular types of organisms are found within an environment.

**Quadrat:** A square frame randomly placed, to estimate number of plants and animals in a given area.



## Feeding Relationships.

**Food Chain:** a series of organisms each dependent on the last as a source of food.

**Food webs:** many food chains linked together to show the feeding relationships of organisms in an ecosystem.

**Producer:** A plant (or photosynthesizing microbe) can make its own food (glucose) using photosynthesis.

**Consumer:** An organism that obtains its food by feeding off of other organisms.

**Trophic level:** the position an organism occupies in a food web, shown by the number of steps it is from the start of the chain.

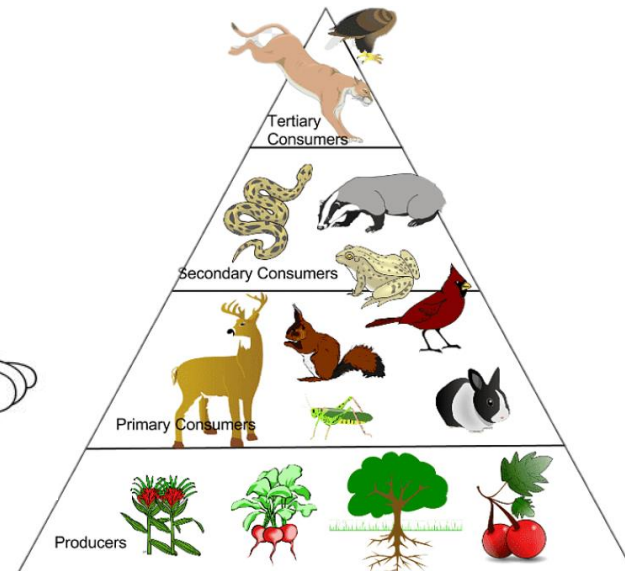
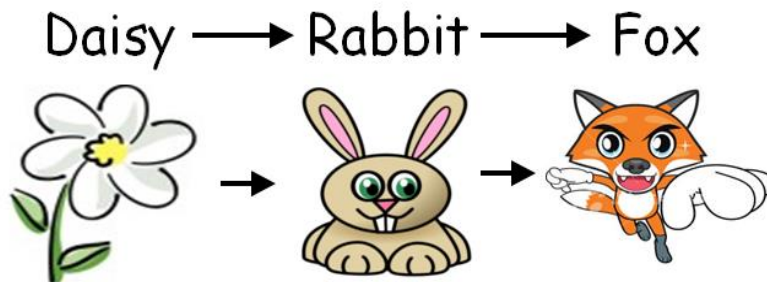
**Pyramid of numbers:** is a graphical representation that shows the number of organisms at each trophic level.

**Biomass:** the total quantity (kg) of organisms in a given area or volume.

**Pyramid of biomass:** A graphical representation of the amount of organic material found in a particular habitat at ascending trophic levels of a food chain.

Food Chain.

\* → Shows the movement of energy.



Keyword	Definition
<b>Habitat</b>	The area in which an organism lives
<b>Ecosystem</b>	The interaction between plants , animals, and their habitats in a particular location
<b>Community</b>	The collection of different types of organisms present in an ecosystem
<b>Adaptation</b>	Characteristics that help an organism to survive in its environment
<b>Structural adaptation</b>	Physical feature that the animal has to help it survive
<b>Behavioural adaptation</b>	Something the animal does to aid survival e.g. migration , hibernation
<b>Extremophile</b>	An organism that can survive and reproduce in extreme conditions

**Habitats** are places where organisms live.

Examples of habitats include:

Desert, meadow, woodland, grassland forest, seashore, ocean.



Living things are adapted to their habitats. Animals and plants have special **adaptations** or characteristics that help them survive in the habitats.



An African elephant, for example, lives in a hot habitat and has very large ears that it flaps to keep cool.



An Arctic fox lives in a cold habitat, it has thick fur to keep it warm.

Large ears and thick fur are examples of **structural adaptations**.